

RNI No. RAJHIN/2007/26996
ISSN 0976-7452 Raaso
Refreed Journal

रासो

अन्तर्विषयी त्रैमासिक शोध पत्रिका

वर्ष-17, अंक-2, मार्च अप्रैल मई 2024





RNI No. RAJHIN/2007/26996
ISSN 0976-7452Raaso
Refreed Journal

रासो

इतिहास, संस्कृति, साहित्य एवं दर्शन की त्रैमासिक शोध पत्रिका

वर्ष-17

अंक - 2

मार्च-अप्रैल-मई-2024

सम्पादक

डॉ. दिनेश चन्द्र शर्मा

निदेशक

कॉम्पिटिशन प्लस संस्थान, ग्रुप ऑफ इन्स्टीट्यूशन्स, अजमेर (राजस्थान)

संरक्षक

* प्रो. जे. पी. शर्मा - पूर्व कुलपति, मोहनलाल सुखाड़िया विश्वविद्यालय, उदयपुर (राजस्थान)

सलाहकार मण्डल

- * प्रो. विभा उपाध्याय - प्रोफेसर, इतिहास एवं भारतीय संस्कृति विभाग, राजस्थान विश्वविद्यालय, जयपुर (राज.)
- * नर्मदा प्रसाद उपाध्याय - प्रसिद्ध साहित्यकार तथा संस्कृति चिंतक, इन्दौर (मध्य प्रदेश)
- * प्रो. टी.के. माथुर - पूर्व अध्यक्ष, इतिहास विभाग महर्षि दयानन्द सरस्वती विश्वविद्यालय, अजमेर (राज.)
- * डॉ. रमेश अग्रवाल - पूर्व स्थानीय सम्पादक, दैनिक भास्कर, अजमेर (राजस्थान)
- * डॉ. गौरव बिस्सा - सहआचार्य, विभागाध्यक्ष, प्रबंध अध्ययन विभाग, राजकीय अभियांत्रिकी महाविद्यालय, बीकानेर (राज.)
- * डॉ. बीना शर्मा - पूर्व विभागाध्यक्ष, हिन्दी विभाग, सावित्री कन्या महाविद्यालय, अजमेर
- * प्रो. शोभा आर. मिश्रा - आचार्य एवं विभागाध्यक्ष, दर्शनशास्त्र विभाग, माधव महाविद्यालय, उज्जैन
- * डॉ. वेद प्रकाश विद्यार्थी - पूर्व सहायक निदेशक, केन्द्रीय हिन्दी निदेशालय, नई दिल्ली।

☆ उप सम्पादक

डॉ. रीता पारीक

प्राचार्य

हुकुमचन्द नोबल इंस्टीट्यूट ऑफ साइंस एण्ड टेक्नोलॉजी, अजमेर

☆ प्रबन्ध सम्पादक

डॉ. मृत्युंजय शर्मा

सह-आचार्य

हुकुमचन्द नोबल इंस्टीट्यूट ऑफ साइंस एण्ड टेक्नोलॉजी, अजमेर

☆ सहायक सम्पादक

डॉ. निष्काम शर्मा

☆ शब्द संयोजन

मुकेश कुमार माहेश्वरी

☆ सम्पादकीय/सचिव

व्यवस्थापकीय कार्यालय

सृजन संस्थान,

आचमन, वेद विहार,

लोहाखान, अजमेर-06

दूरभाष : 0145-2426610

e-mail : rasoajmjournal@gmail.com

☆ प्रकाशक :

पुष्पा शर्मा

सृजन संस्थान,

द्वारा आचमन पब्लिकेशन्स,

अजमेर

☆ मुद्रक :

मैसर्स आर्य प्रिन्टर्स,

केसरगंज, अजमेर

☆ कम्प्यूटराइज्ड सेटिंग :

श्रीनिवास प्रिन्टोग्राफिक्स

132, पंचवटी कॉलोनी,

अजमेर दूरभाष : 0145-2442701

सहयोग दर

* वार्षिक सदस्यता - 500 रु. (चार अंक)

संस्थाओं हेतु - 600 रु. (चार अंक)

कृपया बैंक/ड्राफ्ट/मनीऑर्डर श्रीमती पुष्पा शर्मा, प्रकाशक, आचमन पब्लिकेशन्स,
अजमेर- 305006 के नाम भेजें।

प्रकाशक-श्रीमती पुष्पा शर्मा, 51/1509, वेद विहार, लोहाखान, अजमेर द्वारा मैसर्स सृजन संस्थान, अजमेर
की ओर से प्रकाशित एवं मुद्रक-श्रीमती उषा गर्ग द्वारा मैसर्स आर्य प्रिन्टर्स, केसरगंज, अजमेर से मुद्रित।

Biomass Conversion into Green Chemicals and Biofuels: A Sustainable Approach for the Future

Kripa Sharma

*Asstt. Professor (Department of Science & Technology)
Hukumchand Noble Institute of Science & Technology, Ajmer*

1. Introduction

A rapidly growing world population and increasing energy demand have intensified the search for sustainable alternatives to fossil fuels. Biomass-organic material derived from plants, microorganisms, and agricultural residues-offers a promising renewable feedstock that can be converted into green chemicals, biofuels, and bioenergy. Biomass is considered carbon-neutral because the CO₂ released during its conversion is roughly equal to that absorbed during plant growth.

Chemical conversion of biomass uses principles of green chemistry, catalysis, biotechnology, and thermochemical transformations to produce cleaner, non-toxic, and biodegradable products. In the era of climate policies such as the Paris Agreement and the UN Sustainable Development Goals (SDGs), biomass conversion is gaining global significance.

2. Types and Sources of Biomass

Biomass resources used in chemical and fuel production are categorized as:

2.1 Lignocellulosic Biomass

The most abundant non-food source including:

Agricultural waste (rice husk, wheat straw, sugarcane bagasse), Forestry residue, Wood chips ,Composed of cellulose, hemicellulose, and lignin, which can be chemically broken down.

2.2 Algal Biomass

Microalgae and macroalgae containing high lipid content. Ideal for biodiesel and high-value chemicals.

2.3 Industrial and Municipal Waste

Food waste, Paper waste, Sewage sludge

शोध-पत्र में उल्लिखित सन्दर्भ एवं विचारों के लिए लेखक स्वयं उत्तरदायी है, सम्पादक नहीं।

2.4 Energy Crops : Specially grown plants

Switchgrass, Miscanthus, Jatropha Rich in oils, carbohydrates, and lignin.

3. Chemistry of Biomass Components

Biomass primarily contains:

3.1 Cellulose: A linear polymer of glucose $(C_6H_{10}O_5)_n$. Hydrolyzed to glucose → fermented to ethanol, butanol, and organic acids.

3.2 Hemicellulose: A branched polymer of pentoses (xylose, arabinose) and hexoses. Converted into furfural, xylitol, ethanol.

3.3 Lignin: A highly aromatic polymer. Converted to phenolic compounds, bio-oils, resins, and carbon materials.

3.4 Extractives & Lipids : Essential oils, fatty acids, and triglycerides. Ideal for biodiesel production via transesterification.

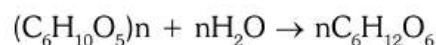
4. Biomass Conversion Technologies

Biomass can be converted through biochemical, thermochemical, and catalytic pathways.

4.1 Biochemical Conversion

4.1.1 Enzymatic Hydrolysis: Cellulases and hemicellulases break down polysaccharides into fermentable sugars.

Reaction:



4.1.2 Fermentation

Microorganisms convert sugars to biofuels and chemicals.

Products:

Bioethanol (*Saccharomyces cerevisiae*)

Biobutanol (*Clostridium acetobutylicum*)

Lactic acid (*Lactobacillus*)

Succinic acid

4.1.3 Anaerobic Digestion

Organic matter → Biogas ($CH_4 + CO_2$)

Used for electricity and heat.

4.1.4 Algal Cultivation & Extraction

Algae are grown in photobioreactors; lipids extracted for biodiesel.

4.2 Thermochemical Conversion

4.2.1 Pyrolysis - Heating biomass at 350-650°C without oxygen to

produce:

Bio-oil

Biochar

Syngas

Fast pyrolysis produces bio-oil, which can be upgraded into green chemicals.

4.2.2 Gasification

High temperature (700-1200°C) partial oxidation H_2). syngas

Syngas methanol, hydrogen, Fischer-Tropsch fuels.

4.2.3 Combustion

Direct burning of biomass for heat and power.

4.3 Catalytic Conversion & Green Chemistry

Catalysts convert biomass-derived molecules into valuable chemicals.

4.3.1 Transesterification of Lipids

Vegetable oils + Methanol $\rightarrow$ Biodiesel (Fatty Acid Methyl Esters)

Uses catalysts like: NaOH, KOH, Enzyme lipases

4.3.2 Dehydration Reactions

Glucose $\rightarrow$ 5-HMF (5-Hydroxymethylfurfural), Xylose Furfural. Used in bioplastics and resins.

4.3.3 Hydrogenolysis & Hydrodeoxygenation

Bio-oil upgrading into hydrocarbons.

5. Green Chemicals from Biomass

Biomass-derived chemicals are safer and biodegradable. Major products include:

5.1 Platform Chemicals

The U.S. Department of Energy (DOE) identifies 12 key platform chemicals such as: 5-HMF

Levulinic acid, Succinic acid, Xylitol, Glycerol

These serve as building blocks for polymers, solvents, and pharmaceuticals.

5.2 Biopolymers

Polylactic acid (PLA)

Polyhydroxyalkanoates (PHA)

Starch-based plastics

Used in packaging, textiles, and biomedical applications.

5.3 Bio-based Solvents

Green solvents obtained from biomass include:

Ethyl lactate, Bio-butanol, Terpenes

Used in paints, coatings, pharmaceutical formulations.

6. Biofuels from Biomass

6.1 First-Generation Biofuels

From edible crops (sugarcane, corn). Not fully sustainable due to food-fuel conflict.

6.2 Second-Generation Biofuels

From lignocellulosic biomass.

Includes: Cellulosic ethanol, Fischer-Tropsch diesel, Biohydrogen

6.3 Third-Generation Biofuels

From algae.: High productivity; suitable for aviation fuel.

6.4 Fourth-Generation Biofuels

Genetically engineered microorganisms for direct fuel production.

7. Environmental Benefits

7.1 Reduced Greenhouse Gas Emissions

Biofuels reduce CO₂ by 40-80% compared to fossil fuels.

7.2 Waste Utilization : Agricultural waste products → valuable

Reduces landfill burden.

7.3 Biodegradability

Green chemicals degrade naturally, reducing soil and water pollution.

7.4 Energy Security

Lowers dependence on imported fossil fuels.

8. Challenges in Biomass Conversion

8.1 Recalcitrance of Lignocellulose

Complex structure requires expensive pretreatment.

8.2 High Production Cost

Enzymes, catalysts, and bioreactors raise costs.

8.3 Competition with Food Crops

First-generation biofuels raise food prices.

8.4 Technological Limitations

Scaling up from lab to industry is difficult.

8.5 Land & Water Use : Energy crops require resources, though less than fossil extraction.

Hydrothermal Liquefaction (HTL)

Converts wet biomass ? biocrude Highly efficient for algae and sewage sludge.

8.6 Nanocatalysts

Metal nanoparticles (Ni, Cu, Pt) improve reaction rates and selectivity.

8.7 Integrated Biorefineries

Zero-waste systems producing multiple products:

Fuels, Chemicals, Heat, Fertilizer (biochar)

9. Conclusion

Biomass conversion into green chemicals and biofuels represents a transformative solution for sustainable development. Through biochemical, thermochemical, and catalytic approaches, biomass can be efficiently converted into renewable fuels and eco-friendly chemicals. While challenges such as high cost and complex pretreatments remain, recent innovations in biotechnology, catalysis, and process engineering have significantly improved the efficiency and feasibility of conversion technologies. With ongoing research and supportive policies, biomass-based industries are poised to reshape the global energy and chemical landscape, providing cleaner and greener alternatives to fossil fuels.



References

1. Zhang, Z., & Huber, G. (2024). *Catalytic Conversion of Biomass to Biofuels and Chemicals*. *Chemical Reviews*.
2. U.S. Department of Energy (DOE). (2024). "Top Value-Added Chemicals from Biomass."
3. Mohan, D., Pittman, C. U., & Steele, P. (2023). "Pyrolysis of Biomass for Fuels and Chemicals." *Energy & Fuels*.
4. Li, C., Zhao, X., Wang, A., & Zhang, T. (2024). "Biomass Upgrading via Hydrogenolysis and Hydrodeoxygenation." *Green Chemistry*.
5. Chisti, Y. (2025). "Algae-Based Biofuels: Progress and Challenges." *Biotechnology Advances*.
6. European Biorefinery Roadmap (2024). "Future of Sustainable Biomass Conversion Technologies."